

## Chemistry Midterm Study Guide

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### Chemistry Midterm Study Guide

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Chemistry Midterm Study Guide Scientific Method: Experimental Variables: o Dependent: depends on what happened in the experiment (y-axis) o Independent: factor that can be manipulated (x-axis) o Constant: does not change in the experiment o Control: group that did not receive treatment Hypothesis: relationship between 2 variables Theory: EXPLAINS something based on observations; may change (Ex ...

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(look on study guide) -The atomic number of this atom is 2, based on the number of protons -The mass number for this atom is the sum of the protons and electrons -This is a +1 charged ion because it has fewer electrons than when it's neutral

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Study Guide for Chemistry 120 Midterm (questions with selected answers) Chemistry 120 – Fall 2020 Selected answers are in blue and italics. Note that I have not double checked each answer so IF you do find a mistake just let me know! This is meant to serve as a study guide and takes material from the first four weeks of the class that is supportive of the Course Outcomes given in the Syllabus.

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A graduated cylinder contains 44.2 milliliters of water. a 48.6-g piece of metal is carefully dropped into the cylinder. when the middle is completely covered with water the water rises to the 51.3 milliliter mark. what is the density of the metal? 6.8 g/mL

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### Midterm - Mrs. Furr's Website

Chemistry 1027b Midterm Study Guide Matter Properties of matter; physical and chemical (what is the difference?) A physical property is one that can be observed without any change in composition. A chemical property is one that is observable in chemical reactions and changes the identity of one or more reactants.

### Chemistry Midterm Guide - Chemistry 1027b Midterm Study ...

Chemistry Midterm Study Guide. Just from \$13.9/Page. Get custom paper. 02\*10^23, Mass number (Z) is the sum of protons plus neutrons •Isotopes of an element have the same number of protons, but different numbers of neutrons. •Isotopes of an element have the same number of protons, but different numbers of neutrons.

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### McCarthy, Mrs. / Chemistry Midterm Study Guide Key

Midterm Exam 2 Study Guide: CHEM-3-94926 - Introductory Inorganic Chemistry Midterm Exam 2 Study Guide The second midterm exam will cover Lectures 8-13 & Chapters 6-8 in the textbook. See below for a comprehensive summary of the topics included on the exam.

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