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a neutron is emitted
from the nucleus. 1_0n .

Particle for Neutron

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Examples • Prob: -An atom of Kr has a mass of 94 AMU. How many protons & neutrons does it have? •

Answer: Kr = element #36 (36 protons).

Neutrons = mass - protons, = $94 - 36 = 58$ neutrons. • Prob: An

atom has 20 protons and 22 neutrons. Draw the

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The production of

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energy in a nuclear reactor can be stopped by pulling out all control rods. A breeder reactor produces more fuel than it uses. The fission products produced in nuclear power plants are not radioactive. An uncontrolled chain reaction led to the nuclear accident in Chernobyl, Ukraine.

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list 5 characteristics of
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Nuclear chemistry is the study of reactions that involve changes in nuclear structure. The chapter on atoms, molecules, and ions introduced the basic idea of nuclear structure, that the nucleus of an atom is composed of protons and, with the exception of ${}^1_1\text{H}$, neutrons.

25.1: Radioactivity - Chemistry

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Chapter 25

Nuclear Chemistry 16.1

The Nucleus and

Radioactivity Our

journey into the center

of the atom begins

with a brief review. You

learned in Chapter 3

that the protons and

neutrons in each atom

are found in a tiny,

central nucleus that

measures about

$1/100,000$ the

diameter of the atom

itself. You also learned

Chapter 16 NuClear

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Chapter 25 – Nuclear
Chemistry
Radioactivity

- Radioactivity is the process by which nuclei emit particles and rays as they break down.
- The name of the penetrating rays emitted by a radioactive source is called radiation.
- A radioactive isotope is an unstable atom which breaks down on its own, releasing

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Section 10.1

Radioactivity (pages

292-297) This section

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Chapter 25

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discusses the different types of nuclear radiation and how they affect matter.

Chapter 10 Nuclear Chemistry Section

10.1 Radioactivity ...

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Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry
Marie Curie went on to show that unlike

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chemical reactions,
radioactivity is not
affected by changes in
physical conditions
such as temperature
and pressure She is the
only person in history
to

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